

A Formal Synthesis of (\pm)-Compactin

Shang-Cheng Hung and Chun-Chen Liao*

Department of Chemistry, National Tsing Hua University, Hsinchu, Taiwan 30043, Republic of China

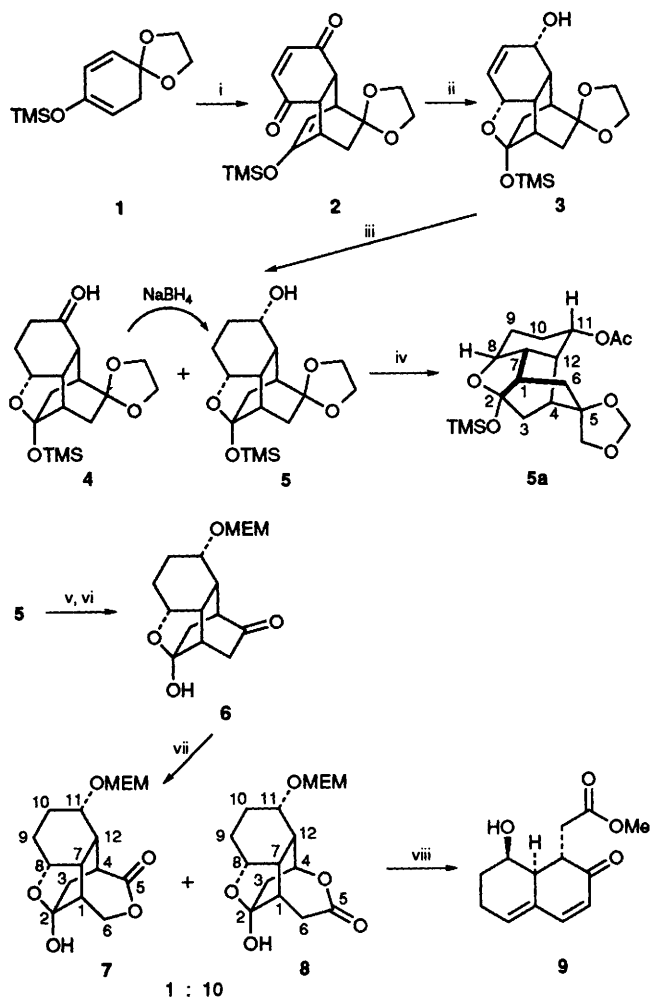
Compound **9**, a key intermediate in the synthesis of (\pm)-compactin, has been prepared from the Diels–Alder adduct of **1** and *p*-benzoquinone in seven steps.

The compactin-mevinolin family,¹ owing to their potent hypocholesterolaemic activity,² have attracted much attention; many strategies of synthesis have been developed.³ Numerous approaches were based on the construction of the hydronaphthalene and lactone portions that were coupled at a

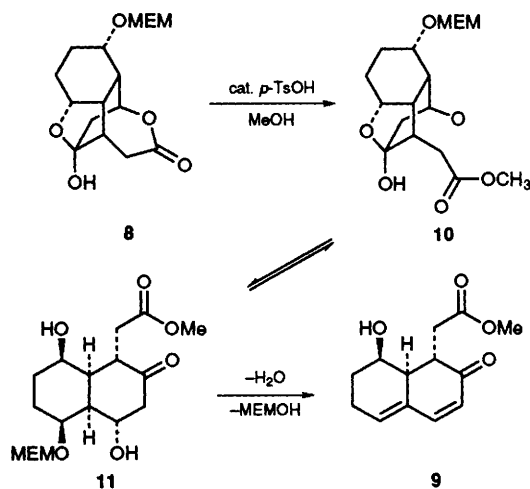
late stage. We report here the synthesis of **9**, which has been reported to be transformed into (\pm)-compactin.⁴

The synthetic sequence is shown in Scheme 1. Compound **2**, prepared from the Diels–Alder reaction of **1**⁵ and *p*-benzoquinone, was transformed into **3** by Luche's reduction.

Catalytic hydrogenation of **3** afforded a mixture of **5** and **4** in 60 and 31% yields respectively; the latter was effectively transformed into **5** by reduction with NaBH₄. The *endo* stereochemistry in **2**, which was expected to be the major



Scheme 1 Reagents and conditions: i, *p*-benzoquinone, PhH, 90 °C, (95%); ii, CeCl₃·7H₂O, NaBH₄, 0 °C, (68%); iii, Ru/C H₂ (92%); iv, AcCl, pyridine; v, MEMCl, Pr₃N⁺Et, (85%); vi, 1 mol dm⁻³ HCl, (98%); vii, *m*-chloroperbenzoic acid, (82%); viii, *p*-TsOH (cat.), MeOH, CH(OMe)₃, 90 °C, (57%) (TMS = trimethylsilyl)



Scheme 2

isomer from theoretical point of view, was mainly deduced from the formation of ketal in **3** after Luche's reduction; this result would not be possible for the *exo* adduct. This result also determined the stereochemistry of C-8 (as indicated in **5a**) in **3-8**. The assignments of the stereochemistry of the hydroxy group in **5** and the corresponding functional groups in **5a-8** were based on the ¹H NMR spectral pattern of the hydrogen atom on C-11 in **5a**, a pattern of ddd (*J* 12.1, 7.0, 4.7 Hz); the large coupling constant 12.1 Hz is due to the coupling between two vicinal diaxial proton indicating that the hydrogen on C-11 is in the axial position. The stereochemistry of the hydroxy group in **3** is expected to be the same as that in **5**. Treatment of **5** with methoxyethoxymethyl chloride (MEMCl) in the presence of ethyldiisopropylamine followed by acidic hydrolysis of the ketal moiety yielded **6**. Baeyer-Villiger oxidation of **6** produced a mixture of **7** and **8** in the ratio 1:10 determined from the integration of the ¹H NMR spectrum [δ 4.39 (*J* 12.2, 3.3 Hz) and δ 4.53 (*J* 12.2, 3.3 Hz) for the two protons on C-6 in **7**; δ 4.12 (*J* 3.1 Hz) for the proton on C-4 in **8**]. Treatment of the mixture of **7** and **8** with toluene-*p*-sulfonic acid (*p*-TsOH) and trimethyl orthoformate in methanol generated **9** in 57% yield after column chromatography. The formation of **9** from **8** is presumably *via* the reaction pathway depicted in Scheme 2. Opening of the lactone ring by transesterification furnished **10** which equilibrated with its keto form **11**. Consecutive elimination of H₂O and MEMOH from **11** yielded the desired product **9**[†] that was converted to (±)-compactin by Girotra and Wendler.⁴

We thank the National Science Council of the Republic of China for financial support.

Received, 12th May 1993; Com. 3/02698A

References

- W. A. Alberts, J. Chen, G. Kuron, V. Hunt, J. Huff, C. Hoffman, J. Rothrock, M. Lopez, H. Joshua, E. Harris, A. Pachett, R. Monaghan, S. Currie, E. Stapley, G. Albers-Schönberg, O. Hensens, J. Hirshfield, K. Hoogsteen, J. Liesch and J. Springer, *Proc. Natl. Acad. Sci. USA*, 1980, **77**, 3957, and references therein.
- A. Endo, *J. Med. Chem.*, 1985, **28**, 401; W. F. Hoffman, A. W. Alberts, P. S. Anderson, J. S. Chen, R. L. Smith and A. K. Willard, *J. Med. Chem.*, 1986, **29**, 849, and references therein.
- D. L. J. Clive, K. S. Keshava Murthy, A. G. H. Wee, J. Siva Prasad, G. V. L. da Silva, M. Majewski, P. C. Anderson, C. F. Evans, R. H. Haugen, L. D. Heerze and J. R. Barrie, *J. Am. Chem. Soc.*, 1990, **112**, 3018, and references therein; A. E. DeCamp, T. R. Verhoeven and I. Shinkai, *J. Org. Chem.*, 1989, **54**, 3207; J. C. Barrish, P. M. Wovkulich, P. C. Tang, A. D. Batcho and M. R. Uskokovic, *Tetrahedron Lett.*, 1990, **31**, 2235; A. R. Daniewski and M. R. Uskokovic, *Tetrahedron Lett.*, 1990, **31**, 5599; S. Hanessian, P. J. Roy, M. Petrini, P. J. Hodges, R. D. Fabio and G. Carganico, *J. Org. Chem.*, 1990, **55**, 5766; A. E. De Camp, S. G. Mills, A. T. Kawaguchi, R. Desmond, R. A. Reamer, L. DiMichele and R. P. Volante, *J. Org. Chem.*, 1991, **56**, 3564.
- N. N. Girotra and N. L. Wendler, *Tetrahedron Lett.*, 1982, **23**, 5501; 1983, **24**, 3687; N. N. Girotra, N. L. Wendler and R. A. Reamer, *Tetrahedron Lett.*, 1984, **25**, 5371.
- S.-C. Hung and C.-C. Liao, *Tetrahedron Lett.*, 1991, **32**, 4011.

[†] Pertinent spectral data of **9**: IR ν /cm⁻¹ (CHCl₃) 3447, 2927, 1728, 1663, 1437, 1209, 1092, 859; ¹H NMR (400 MHz, CDCl₃) δ 6.96 (d, *J* 9.7 Hz, 1 H), 6.22 (t, *J* 2.4 Hz, 1 H), 5.88 (d, *J* 9.7 Hz, 1 H), 4.03 (t, *J* 2.2 Hz, 1 H), 3.70 (s, 3 H), 3.01 (ddd, *J* 13.5, 6.4, 3.9 Hz, 1 H), 2.60 (dd, *J* 13.5, 2.2 Hz, 1 H), 2.46–2.56 (m, 1 H), 2.43 (dd, *J* 17.8, 6.4 Hz, 1 H), 2.25 (m, 1 H), 2.06 (m, 1 H), 1.63 (m, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ 198.45 (C), 175.11 (C), 147.62 (CH), 136.15 (CH), 131.67 (C), 124.46 (CH), 64.28 (CH), 52.22 (CH₃), 44.16 (CH), 43.06 (CH), 31.06 (CH₂), 27.41 (CH₂), 21.33 (CH₂); MS (75 eV) *m/z* 236 (M⁺, 2%), 218 (100%), 186 (76%), 160 (98%), 158 (76%), 145 (76%), 132 (68%); HRMS (EI) Calc. for C₁₃H₁₆O₄ 236.1049, found 236.1054.